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Chapter 11: Chemical Reactions. STUDY. Flashcards. Learn. Write. Spell. Test. PLAY. Match. Gravity. Created by. Demitri_Bautista. Key Concepts: Terms in this set (253) chemical equation ____ is a representation of a chemical reaction with reactants on the left, products on the right, and an arrow separating the two.

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Chemical Reactions Chapter 11 Study Guide (Unit 8) 2 | P a g e The law of conservation of mass states that the mass of the reactants will always equal the mass of the products. In result the number of atoms of one element on the reactants side should be identical to the number of atoms of the same element on the products side.

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Modern Chemistry 71 Chapter Test Name Class Date Chapter Test B, continued 17. A substance combines with oxygen, releasing a large amount of energy as heat and light, in a(n) . 18. The decomposition of a substance by an electric current is called . 19. A(n) orders the elements by the ease with which they undergo certain chemical reactions. 20.

Assessment Chapter Test B - Ed W. Clark High School

COVID-19 Protocols for Public Interaction (Updated as of 7/16/20). Dear Chariho Community, As you are aware, districts across the state are engaged in planning for the reopening of schools in the fall.

Home - Chariho Regional School District

Modern Chemistry 65 Chapter Test Chapter: Chemical Equations and Reactions In the space provided, write the letter of the term or phrase that best completes each statement or best answers each question. _____ 1. You mix solution A with solution B in a beaker. Which of the following observations does not help you prove that a chemical reaction ...

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One sign that a chemical reaction has occurred is the formation of new substances. _____ 17. A chemical equation shows a chemical reaction using symbols instead of words. _____ 18. The reactants are the new materials produced during a chemical reaction. _____ 19. Increasing the surface area of the reactants will speed up a chemical reaction.

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